THE SPECIFIC CHARGE OF THE ELECTRON



· set your system up such that there is no beam deflection:

e = Be v $v = \frac{E}{B} \qquad (A)$

From our knowledge of the electron gun we know that:

 $eV = \frac{1}{2}mV^2$ $\frac{e}{m} = \frac{v^2}{2v}$ By substituting (A) into (B) : $\frac{e}{m} = \frac{\left(\frac{E}{B}\right)^2}{1 \sqrt{2}}$ electric field strength Ē [Vm-'] 6 m 2VB2 specific charge on magnetic field strength electron [7] [Ckg'] accelerating p.d. of electron gun [V] Thomson's conclusions (1897)

1. electrons are indeed particles with mass, charge and momentum

- 2. specific charge of an dectron is 1.7 × 10"Ckg⁻¹ (1800 higher than Hydrogen ian)
- 3. cathode rays are in fact made out of fundamental particles.

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(1) Electron gun produces a beam of e⁻
(2) as the e⁻ pass through the H-gas the collide with the H-atoms causing excitation. H-atoms proceed to quickly de-excite releasing visible photons (allowing e⁻ path to be observed)
(3) Helmholtz coils create a strong magnetic field in bulb
(4) B-field creates a force on electrons (= Bev) that is always perpendicular to motion (due to LHR)
→ acts as a certification force so we can say: mv² = Bev

radius of circular path

Determining Specific charge

Starting from our centripetal force equation :

$$\frac{mv^2}{r} = BeV$$

divide through by V

mv = Be

square result :

$$\frac{m^2v^2}{r^2} = \beta^2 e^{\frac{1}{2}}$$

sub in $\frac{1}{2}mv^2 = eV$:

$$\frac{m(2eV)}{r^2} = B^2 e^{i2}$$

rearrange :

